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HIGH PRESSURE ISOMERIZATION OF 1,4-BIS(METHYLTHIO)HEXAFLUORO-2-BUTENE TO -1-BUTENE

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SUMMARY

The complete bond migration of the internal olefinic bond of trans-1,4-bis(methylthio)hexafluoro-2-butene(I) under high pressure to the terminal olefinic bond to form cis- and trans-1,4-bis(methylthio)hexafluoro-1-butene(II) is described. The Teflon-lined high pressure cell maintains a constant pressure at 16,000 atm and 180 to 200°C for 24 hours. The cis- and trans-II are elucidated from gas chromatography-mass spectroscopy, which shows two identical parent ions at m/e value of 255(C₆H₆F₆S₂⁺), but at different elution times. The ¹⁹F NMR data of the isomeric products are summarized. Elemental analysis of II is verified by double focussing high resolution mass spectrometer.

INTRODUCTION

In the course of our investigation on the methyl disulfide addition reactions to hexafluoro-2-butyne by photolysis [1] and to hexafluoro-butadiene under ultrasonic photolysis [2], we reported the formation of 1:1 adducts (CH₃SSCH₃·C₄F₆). In the former case an equal mole ratio of cis- and trans-2,3-bis(methylthio)hexafluoro-2-butene was formed; whereas in the latter example the main product was the trans-isomer of 1,4-bis(methylthio)hexafluoro-2-butene(I) in the presence of trace amounts of cis-I and other two structurally unidentified isomeric 1:1 adducts. The minor products of the latter example were the isomeric 1:2 adducts (CH₃SSCH₃·2C₄F₆). The oligomeric adducts (e.g., 2CH₃SSCH₃·2C₄F₆, CH₃SSCH₃·3C₄F₆ and CH₃SSCH₃·4C₄F₆) in trace amounts from the latter example were only identified by gas chromatography-mass spectroscopy [3].

RESULTS AND DISCUSSION

In this paper we report that the isolated trans-I fraction underwent complete bond migration under high pressure from internal to terminal olefinic bond to form cis- and trans-bis(methylthio)hexafluoro-1-butene(II):

TABLE 1
 ^{19}F NMR spectral data of cis- and trans-1,4-bis(methylthio)hexafluoro-1-butene

Compound	Relative Intensity	Assignment	Chemical Shift ϕ (ppm from CFC_3)	Coupling Constant J(Hz)
$ \begin{array}{c} \text{SCH}_3 \\ \\ \text{(1)} \text{FC} \\ \\ \text{(2)} \text{CF} \\ \\ \text{(3)} \text{CF}_2 \\ \\ \text{(4)} \text{CF}_2 \\ \\ \text{SCH}_3 \end{array} $	2	(4)	95.9	$J(3,4)=7.5;$ $J(2,4)=5.0;$ $J(1,4)=6.5;$ $J(2,3)=26;$ $J(1,3)=15;$ $J(1,2)=143$
	2	(3)	118.0	
	1	(2)	128.0	
	1	(1)	157.8	
<u>trans</u> -II				
$ \begin{array}{c} \text{SCH}_3 \\ \\ \text{(1)} \text{FC} \\ \\ \text{(2)} \text{FC} \\ \\ \text{(3)} \text{CF}_2 \\ \\ \text{(4)} \text{CF}_2 \\ \\ \text{SCH}_3 \end{array} $	2	(4)	95.5	$J(3,4)=8;$ $J(2,4)=6;$ $J(2,3)=13;$ $J(1,3)=2(?);$ $J(1,2)=5$
	1	(1)	104.2	
	2	(3)	115.2	
	1	(2)	139.5	
<u>cis</u> -II				

Table 1 summarizes the ^{19}F NMR of the trans- and cis-II. The GC-mass spectral data showed identical parent ion peaks at 256($\text{C}_6\text{F}_6\text{H}_6\text{S}_2^+$) and also same mass cracking patterns, except at different elution times.

Mass spectroscopic weight (CEC21-110-B) of trans- and cis-II: Found, 255.9815. Calculated for $\text{C}_6\text{F}_6\text{H}_6\text{S}_2$: 255.9801.

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